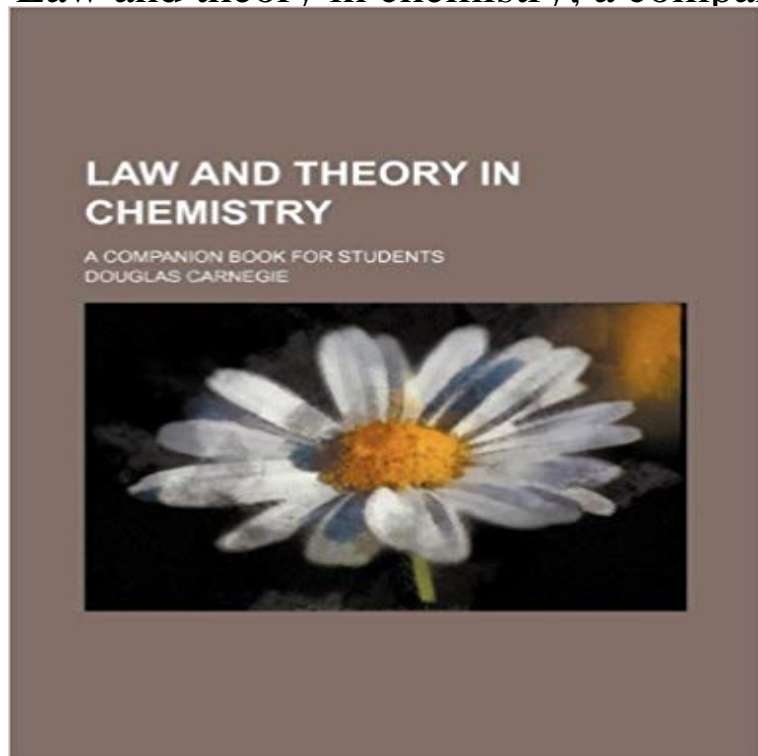


Law and theory in chemistry; a companion book for students



This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1894 Excerpt: ...time (1645-1715) up till the beginning of this century, chemists were possessed of the idea that the products of organisms--most of which are compounds of carbon--could not be made artificially, but that their production absolutely demanded the intermediacy of life--the operation of the so-called vital force. Operations of chymistry fall short of vital force; no chymist can make milk or blood of grass. Further, it was the custom to ascribe the variable results obtained in analyses of animal and vegetable products to the fact that such bodies did not conform to the fundamental law of chemistry (law of fixity of composition), and not to the difficulties attendant on the analytical methods and the purification of complex carbon compounds. Hence a very sharp distinction was made between organic and inorganic or mineral compounds. But the distinction is now recognised as existing for convenience only.¹ Exactly the same laws hold in the case of carbon compounds as in the realm of mineral or inorganic chemistry, and many of those carbon compounds which are products of the life of organisms have been synthesised independently of the living laboratory, i.e., of vital force so called. Thus there exists no better reason than convenience to be adduced for this primary division of compounds, and no valid excuse exists for retaining the effete and misleading term organic compounds in preference to the more fitting title carbon compounds. The compounds of nitrogen are increasing so rapidly, that before long it may be advisable to study them as a class apart, just as has been the case with carbon compounds. In discussing acids, bases, and salts, it will be only too apparent that we

are unable to give satisfactory definitions of these cla...

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